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NEWS 6 JAN 28 USGENE now provides USPTO sequence data within 3 days of publication
NEWS 7 JAN 28 TOXCENTER enhanced with reloaded MEDLINE segment
NEWS 8 JAN 28 MEDLINE and LMEDLINE reloaded with enhancements
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NEWS 12 FEB 25 IMSPRODUCT reloaded with enhancements
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NEWS 14 MAR 31 IFICDB, IFIPAT, and IFIUDB enhanced with new custom IPC display formats
NEWS 15 MAR 31 CAS REGISTRY enhanced with additional experimental spectra
NEWS 16 MAR 31 CA/CAplus and CASREACT patent number format for U.S. applications updated
NEWS 17 MAR 31 LPCI now available as a replacement to LDPCI
NEWS 18 MAR 31 EMBASE, EMBAL, and LEMBASE reloaded with enhancements
NEWS 19 APR 04 STN AnaVist, Version 1, to be discontinued
NEWS 20 APR 15 WPIDS, WPINDEX, and WPIX enhanced with new predefined hit display formats
NEWS 21 APR 28 EMBASE Controlled Term thesaurus enhanced
NEWS 22 APR 28 IMSRESEARCH reloaded with enhancements

NEWS EXPRESS FEBRUARY 08 CURRENT WINDOWS VERSION IS V8.3,
AND CURRENT DISCOVER FILE IS DATED 20 FEBRUARY 2008

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NEWS IPC8 For general information regarding STN implementation of IPC 8

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* * * * * * * * * STN Columbus * * * * * * * * * * * * * * *

FILE 'HOME' ENTERED AT 17:58:52 ON 12 MAY 2008

FILE 'CASREACT' ENTERED AT 17:59:31 ON 12 MAY 2008
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FILE CONTENT:1840 - 10 May 2008 VOL 148 ISS 20

New CAS Information Use Policies, enter HELP USAGETERMS for details.

*
* CASREACT now has more than 13.8 million reactions
*

Some CASREACT records are derived from the ZIC/VINITI database (1974-1999) provided by InfoChem, INPI data prior to 1986, and Biotransformations database compiled under the direction of Professor Dr. Klaus Kieslich.

This file contains CAS Registry Numbers for easy and accurate substance identification.

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L1 STRUCTURE UPLOADED

=>
Uploading C:\Program Files\Stnexp\Queries\ALL.str reactant/reagent

L2 STRUCTURE UPLOADED

=> S L1 FULL
FULL SEARCH INITIATED 18:00:27 FILE 'CASREACT'
SCREENING COMPLETE - 23160 REACTIONS TO VERIFY FROM 5791 DOCUMENTS

100.0% DONE 23160 VERIFIED 4735 HIT RXNS 1392 DOCS
SEARCH TIME: 00.00.01

L3 1392 SEA SSS FUL L1 (4735 REACTIONS)

=> S L3 AND LITHIUM METAL

25160 LITHIUM

51803 METAL

106 LITHIUM METAL

(LITHIUM (W) METAL)

L4 6 L3 AND LITHIUM METAL

=> D L4 IBIB ABS CRD 1-6

L4 ANSWER 1 OF 6 CASREACT COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER: 143:248504 CASREACT
TITLE: Method for producing alkyl lithium compounds and aryl

lithium compounds by monitoring the reaction by means of ir-spectroscopy

INVENTOR(S): Weiss, Wilfried; Dawidowski, Dirk; Pleyer, Walter;

Krueckel, Frank

PATENT ASSIGNEE(S): Chemetall G.m.b.H., Germany

SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

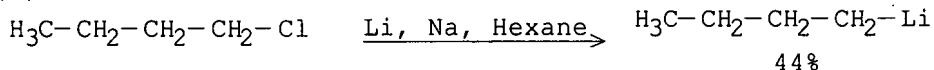
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005082911	A1	20050909	WO 2005-EP1954	20050224
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
DE 102004009445	A1	20050929	DE 2004-10200400944520040227	
EP 1723153	A1	20061122	EP 2005-733858	20050224
R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR				
CN 1922192	A	20070228	CN 2005-80005827	20050224
IN 2006CN03106	A	20070608	IN 2006-CN3106	20060825
US 20070152354	A1	20070705	US 2006-589715	20061023
PRIORITY APPLN. INFO.:			DE 2004-10200400944520040227	
			WO 2005-EP1954	20050224

OTHER SOURCE(S): MARPAT 143:248504

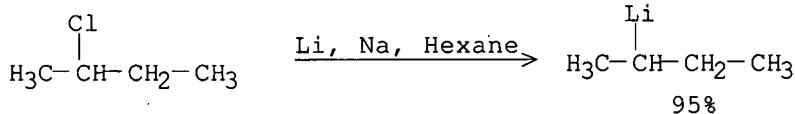
AB The invention relates to a method for producing alkyl lithium compds. and aryl lithium compds. by reacting lithium metal with alkyl or aryl halogenides in a solvent, the concentration of the alkyl/aryl halogenide and the alkyl/aryl lithium compound being detected according to an in-line measurement in the reactor by IR spectroscopy, and an exact recognition of the end point of the dosing of the halogenide constituents being carried out by evaluation of the IR measurement. Said method enables an optimum reactive process and reaction yield. The identification of the resp. concentration of the adduct and the product is a reliable reactive process. The yield of the reaction is also optimized by determining the end point of the halogenide dosing, as is the purity of the product due to a lower concentration thereof during the reaction.

RX(1) OF 5



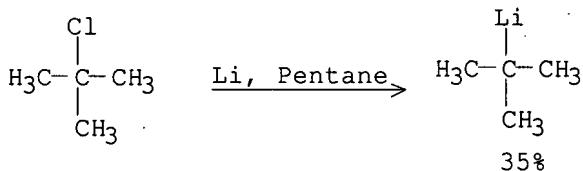
CON: 280 minutes, room temperature

RX(2) OF 5



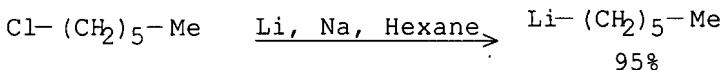
CON: 75 minutes, 40 deg C, 290 atm

RX(3) OF 5



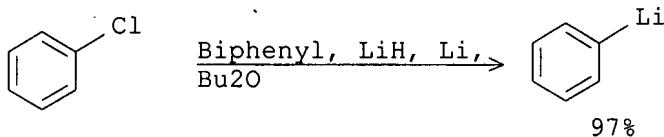
NOTE: *tert*-butyllithium mediated
CON: 144 minutes, room temperature

RX(4) OF 5



CON: 40 deg C, 290 atm

RX(5) OF 5



CON: 4 hours, 35 deg C

REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 2 OF 6 CASREACT COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 142:114226 CASREACT

TITLE: Tetrakis(trimethylsilyl)-substituted cyclobutadiene dianion

dilithium salt: synthesis and structure

AUTHOR(S): Sekiguchi, A.; Matsuo, T.; Tanaka, M.; Watanabe, H.; Nakamoto, M.

CORPORATE SOURCE: Department of Chemistry, University of Tsukuba, Tsukuba, Ibaraki, 305 8571, Japan

SOURCE: Russian Chemical Bulletin (Translation of Izvestiya Akademii Nauk, Seriya Khimicheskaya) (2004), 53(5), 1109-1115

CODEN: RCBUEY; ISSN: 1066-5285

PUBLISHER: Kluwer Academic/Consultants Bureau

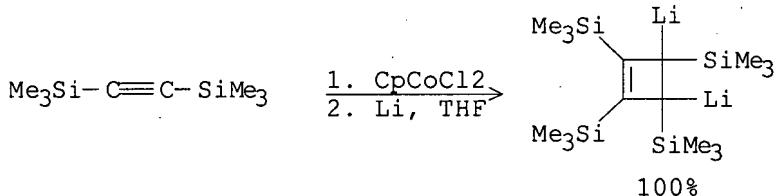
DOCUMENT TYPE: Journal

LANGUAGE: English

AB The reaction of tetrakis(trimethylsilyl)cyclobutadienylcyclopentadienyl cobalt complex $(\text{Me}_3\text{Si})_4\text{C}_4\text{CoCp}$ with lithium metal in THF yielded the dilithium salt of cyclobutadiene dianion CBD_2^- stabilized by four trimethylsilyl groups, $\text{Li}^+2[(\text{Me}_3\text{Si})_4\text{C}_4]_2^-$. The bridged CBD_2^- dianion was also synthesized by a similar procedure starting from the bridged cobalt complex, which was prepared from the reaction of 2,2,5,5,8,8,11,11-octamethyl-2,5,8,11-tetrasilacyclododeca-1,6-diyne with $\text{CpCo}(\text{CO})_2$ in refluxing octane. The aromaticity of the CBD_2^- is discussed on the basis of the structural characteristics and magnetic properties.

RX(1) OF 6 - REACTION DIAGRAM NOT AVAILABLE

RX(5) OF 6 - 2 STEPS



CON: STEP(1) 5 days, reflux
STEP(2) 24 hours, room temperature

REFERENCE COUNT: 65 THERE ARE 65 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 3 OF 6 CASREACT COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 136:232391 CASREACT

TITLE: Chemical process and plant for n-butyl lithium manufacture

INVENTOR(S): Buckley, Glyn Jeffrey; Stairmand, John William; Bowe, Michael Joseph

PATENT ASSIGNEE(S): Accentus PLC, UK

SOURCE: PCT Int. Appl., 15 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

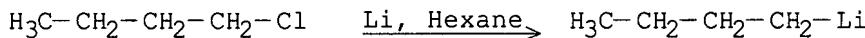
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2002020151	A1	20020314	WO 2001-GB3982	20010905
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW				
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
AU 2001084277	A	20020322	AU 2001-84277	20010905
EP 1320413	A1	20030625	EP 2001-963247	20010905
EP 1320413	B1	20060405		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR				
JP 2004508171	T	20040318	JP 2002-524623	20010905
US 20030168330	A1	20030911	US 2003-343786	20030204
US 6841095	B2	20050111		
PRIORITY APPLN. INFO.:			GB 2000-22016	20000908
			WO 2001-GB3982	20010905

AB A chemical plant for performing a chemical reaction between particles of a material such as lithium metal, and a reagent such as Bu chloride in solution in hexane, in which one reaction product is a solid material, includes a reaction vessel. Several ultrasonic transducers are attached to a wall of the vessel to irradiate ultrasonic waves into the vessel, the vessel being large enough that each transducer irradiates into fluid at least 0.1 m thick, each transducer irradiating no >3 W/cm², and the transducers being sufficiently close to each other and the number of transducers being sufficiently high that the power dissipation within the vessel is at least 10 W/L but no >200 W/L. The high intensity of ultrasound ensures that lithium chloride is cleaned off the surface of the lithium metal particles throughout the vessel.

RX(1) OF 1



NOTE: ultrasound, industrial scale, ultrasound is used to break the byproduct lithium chloride off of the lithium metal

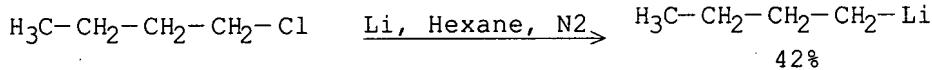
REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 4 OF 6 CASREACT COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER: 129:149087 CASREACT
TITLE: Preparation of alkylolithiums
INVENTOR(S): Iwao, Tetsuya; Yamamura, Kiyoshi
PATENT ASSIGNEE(S): Mitsui Petrochemical Industries, Ltd., Japan; Mitsui Chemicals Inc.
SOURCE: Jpn. Kokai Tokkyo Koho, 3 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 10182658	A	19980707	JP 1996-345795	19961225
JP 3570835	B2	20040929		

PRIORITY APPLN. INFO.: JP 1996-345795 19961225
AB Alkylolithiums are prepared by reaction of alkyl halides with Li containing ≤ 500 ppm N. BuCl was reacted with Li containing 160 ppm N in hexane at room temperature for 30-40 min, then filtered for 1 min to give 42% BuLi.

RX(1) OF 1



NOTE: room temp. 30-40 min

L4 ANSWER 5 OF 6 CASREACT COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER: 123:112396 CASREACT
TITLE: Process of preparing trimethylsilyloxy functionalized alkylolithium compounds
INVENTOR(S): Schwindeman, James A.
PATENT ASSIGNEE(S): FMC Corp., USA
SOURCE: U.S., 5 pp.
CODEN: USXXAM
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5403946	A	19950404	US 1994-279721	19940725
US 5543540	A	19960806	US 1994-341822	19941121
WO 9603408	A1	19960208	WO 1995-US9256	19950724

W: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI,
GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LT, LU, LV, MD,
MG, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ,

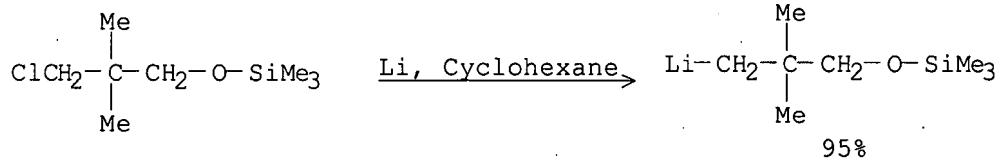
TT, UA
 RW: KE, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT,
 LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE,
 SN, TD, TG
 AU 9531410 A 19960222 AU 1995-31410 19950724
 EP 800525 A1 19971015 EP 1995-927358 19950724
 EP 800525 B1 20030409
 R: DE, FR, GB, NL
 JP 10504813 T 19980512 JP 1996-505889 19950724
 US 5912378 A 19990615 US 1997-851324 19970505
 PRIORITY APPLN. INFO.: US 1994-279721 19940725
 US 1994-341822 19941121
 WO 1995-US9256 19950724
 US 1996-637192 19960408

OTHER SOURCE(S): MARPAT 123:112396

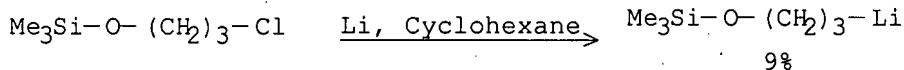
AB A process for producing compds. of the formula Me_3SiORLi ($\text{R} = \text{C}_2\text{-10 alkyl, C}_6\text{-10 aryl}$) by reacting haloalc. HORX ($\text{R} = \text{same, X} = \text{Cl, Br}$) with hexamethyldisilazane, in an inert atmospheric in hydrocarbon solvent, at a temperature

between 20° and reflux temperature of the solvent followed by lithiation with powdered lithium metal, is described. Thus, reaction of 3-chloro-2,2-dimethyl-1-propanol with hexamethyldisilazane in cyclohexane gave 3-chloro-2,2-dimethyl-1-trimethylsiloxypropane which on lithiation with lithium dispersion gave title compound, 3-chloro-2,2-dimethyl-1-trimethylsiloxypropyllithium.

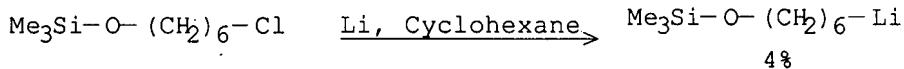
RX(4) OF 9



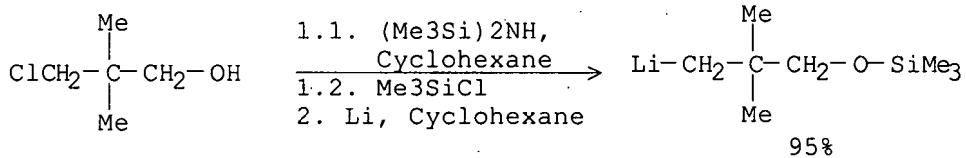
RX(5) OF 9



RX(6) OF 9

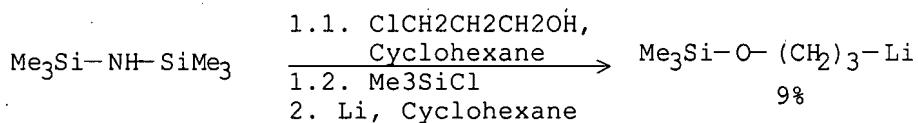


RX(7) OF 9 - 2 STEPS



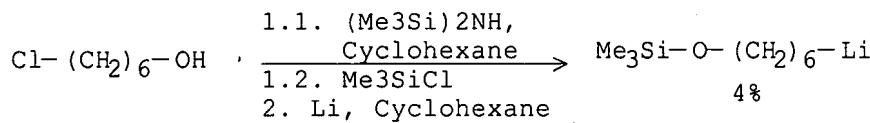
NOTE: 1) TMS-CL ADDED IN TWO BATCHES WITH HEATING BETWEEN ADDITIONS

RX(8) OF 9 - 2 STEPS



NOTE: 1) TMS-CL ADDED IN TWO BATCHES WITH HEATING BETWEEN ADDITIONS

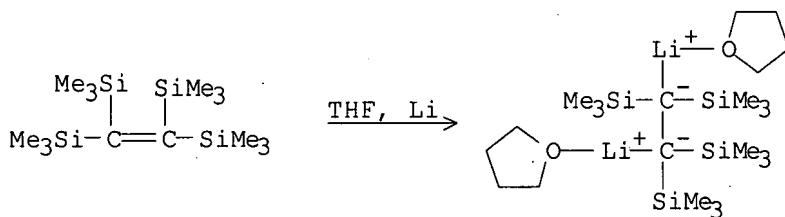
RX (9) OF 9 - 2 STEPS



NOTE: 1) TMS-CL ADDED IN TWO BATCHES WITH HEATING BETWEEN ADDITIONS

L4 ANSWER 6 OF 6 CASREACT COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER: 110:212879 CASREACT
TITLE: Isolation and characterization of 1,2-dilithio[tetrakis(trimethylsilyl)]ethane. The first crystal structure of nonconjugated 1,2-dilithioethane
AUTHOR(S): Sekiguchi, Akira; Nakanishi, Tetsuo; Kabuto, Chizuko; Sakurai, Hideki
CORPORATE SOURCE: Fac. Sci., Tohoku Univ., Sendai, 980, Japan
SOURCE: Journal of the American Chemical Society (1989), 111(10), 3748-50
CODEN: JACSAT; ISSN: 0002-7863
DOCUMENT TYPE: Journal
LANGUAGE: English
AB Careful reduction of tetrakis(trimethylsilyl)ethylene with lithium metal in THF gave the a very hygroscopic and air sensitive title compound (I), the first nonconjugated 1,2-dilithioethane derivative. NMR spectra, chemical reactions, and x-ray crystallog. data of I are described.

RX(1) OF 8



=>

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DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-4.50	-4.50

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